standing of bonding in these and other solid-state systems. But if compounds have not been prepared, their physical properties cannot be measured! The present eclectic compilation of novel and interesting structure types underscores the enormous potential of the reactive flux method for the synthesis of new solid-state materials.

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Cytochalasan Synthesis: Macrocycle Formation via **Intramolecular Diels-Alder Reactions**

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The cytochalasans are fungal metabolites of considerable importance because of their potent biological activities, including effects on transport across mammalian cell membranes and cell morphology.^{1,2} Structurally they are characterized by the presence of a reduced isoindolone nucleus fused to a macrocyclic ring, which can be either a lactone, as in cytochalasin B (1)² a carbonate, as in cytochalasin E (2)³ or a carbocycle, as in cytochalasins D^{2} , H^{4} , and K (3-5).⁵ The biosynthesis of the cytochalasans has been studied⁶ and the N(2)-C(4) fragment shown to be derived from an amino acid, usually phenylalanine,⁷ although other amino acids can be incorporated, e.g., tryptophan into cytochalasin G $(6)^8$ and leucine into the aspochalasans, e.g., aspochalasin C $(7)^9$ (Scheme I).

In developing a synthesis of the cytochalasans, two problems that have to be overcome are the control of stereochemistry, particularly around the heavily functionalized isoindolone nucleus, and the efficient formation of the macrocyclic ring. Most approaches to the isoindolone have used Diels-Alder reactions. However, the efficient formation of macrocyclic rings is still a major problem in organic synthesis because of competing intermolecular processes, and several approaches have been developed for the introduction of the macrocyclic ring into cytochalasan precursors including ring-expansion and fragmentation processes, together with direct ring formation. The simultaneous formation of the isoindolone and the large ring via an intramolecular Diels-Alder reaction is the focus of the present Account, although relevant intermolecular Diels-Alder reactions will be discussed first to introduce the chemistry that is involved.

Hydroisoindolone Synthesis via Intermolecular **Diels-Alder Reactions**

Synthesis of the isoindolone fragment of the cytochalasans via Diels-Alder reactions of pyrrol-2(5H)-ones

Scheme II Pyrrol-2(5H)-ones as Dienophiles a 10 CH₂Ph 12 150-200 15 $13 R^1 = H, R^2 = CH_2CI$ $\mathbf{14} \mathbf{R}^1 = n-Bu$, $\mathbf{R}^2 = Ph$, Me, OCH₂Ph has been widely investigated, with the nature of the N-substituent being found to be particularly important. (1) Pendse, G. S. Recent Advances in Cytochalasans; Chapman and

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Stereoselectivity of diene approach to pyrrol-2-Figure 1. (5H)-ones.

The parent pyrrol-2(5H)-one (8)¹⁰ is thermally unstable, and the 3-acetylpyrrol-2(5H)-one 9 undergoes irreversible tautomerism to its enol 10,¹¹ and so these are unsuitable for use as dienophiles. Moreover, the 1,5dibenzyl-3-butylpyrrol-2(5H)-one (11) isomerizes to its 5-benzylidene isomer 12 rather than undergoing Diels-Alder reaction with (E,E)-2,4-hexadiene or cyclopentadiene.¹² However, the introduction of an electron-withdrawing substituent at N(1) dramatically changes the chemistry of pyrrol-2(5H)-ones, and they become useful Diels-Alder dienophiles, e.g., the Nacylpyrrol-2(5H)-ones 13¹³ and 14¹² both react stereoselectively with (E,E)-2,4-hexadiene (Scheme II).

The difference in reactivity between the N-acyl- and N-alkylpyrrol-2(5H)-ones has been attributed to an increase in the antiaromatic character of the enol tautomer of the N-acyl compounds due to delocalization of the N lone pair of electrons into the exocyclic carbonyl group.¹⁴ This phenonemon also prevents racemization of the 1-acyl-5-benzylpyrrol-2(5H)-ones during **Diels-Alder reactions.**

Of particular interest in connection with cytochalasan synthesis is the stereoselectivity of these reactions. It appears that there is a preference for the diene to approach the less hindered face of the dienophile, away from the benzyl substituent at C(5), endo to the pyrrol-2(5H)-one ring (see Figure 1). This mode of addition gives rise to the stereochemistry required for a cytochalasan synthesis at all five stereogenic centers and is a key feature of this synthetic approach.

Rather vigorous conditions were required for Diels-Alder addition of the monoactivated pyrrol-2(5H)-ones 13 and 14 to conjugated dienes. However, N-acylpyrrol-2(5H)-ones with electron-withdrawing groups at the 3-position, being doubly activated dienophiles, are more reactive, although somewhat unstable, and undergo Diels-Alder reactions with a wider range of conjugated dienes and trienes. For example, the 3-(ethoxycarbonyl)pyrrol-2(5H)-one 16 reacts with triene 17

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to give adduct 18,¹⁴ and the 3-acylpyrrol-2(5H)-one 19 reacts with (E,E)-2,4-hexadiene to give a mixture of adducts 20 and 21.¹² Again the major product of these reactions was formed by addition of the diene to the less hindered face of the dienophile endo to the pyrrol-2-(5H)-one ring (Figure 1; Scheme III).

An intermolecular Diels-Alder reaction was used in the first synthesis of cytochalasin B (1),¹⁵ in which the triene 22 and the monoactivated 3-acetoxypyrrol-2-(5H)-one 23 were combined to give adduct 24 (170 °C, 4 days, 40% conversion). Conversion into the seco acid 25, which had previously been transformed into cytochalasin B (1),¹⁶ completed the total synthesis (Scheme IV).

Simultaneous Macrocycle and Hydroisoindolone Formation by Intramolecular **Diels-Alder Cyclization**

The work outlined above demonstrated that Diels-Alder reactions can be used to synthesize the isoindolone fragment of the cytochalasans. However, there remained the difficulty of assembling the macrocyclic ring. One attractive approach appeared to be to use the Diels-Alder reaction in an *intra*molecular mode to assemble the isoindolone and the macrocyclic ring simultaneously. This strategy would avoid complications of regioselectivity encountered in an intermolecular Diels-Alder approach, but it was not clear whether polymerization of the diene-dienophile via intermolecular cycloaddition reactions would interfere.

Formation of the large-ring system of the cytochalasans via an intramolecular Diels-Alder reaction was first investigated using diene anhydride 26. Heating a dilute solution of 26 in toluene under reflux induced cyclization to give the adducts 27 and 28, so demonstrating

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the viability of this approach. However, the yield was only modest because of competing hydrolysis of the enol ester under the high dilution conditions.^{17,18} Nevertheless, this strategy was used successfully in the second synthesis of cytochalasin B(1), in which the trienyl pyrrol-2(5H)-one 29 was heated at 180-190 °C for 6 days to give a mixture of Diels-Alder products, yield 30%, ratio ca. 4:1, with the required isomer 30 as the major component. Introduction of the 21,22-double bond and oxidation at C(6)-C(7) completed the synthesis of cytochalasin B^{19} (Scheme V).

The greater reactivity of the doubly activated 3acylpyrrol-2(5H)-ones, e.g., 19, as dienophiles, suggested that intramolecular Diels-Alder reactions of these compounds would provide useful access to carbocyclic cytochalasans. Difficult steps in the synthesis were identified as the stereoselective synthesis of the conjugated triene, the assembly of the sensitive 3-acylpyrrol-2(5H)-one incorporating the conjugated triene. and the Diels-Alder cyclization itself. Procedures for carrying out these reactions were developed during a synthesis of proxiphomin (39)^{20,21} (Scheme VI).

The dienvl phosphonate 31²⁰ was condensed with the homochiral aldehyde 32 to give the acid-sensitive (E, -E,E)-trienyl ester 33 (60%), only minor amounts of other stereoisomers being detected, and the ester was converted into the acyl imidazolide 34. The homochiral pyrrolidinone 42, available from (S)-phenylalanine, was deprotonated by using lithium hexamethyldisilazide, and the enolate so obtained was acylated by using imidazolide 34 to give the 3-acylpyrrolidinone 35. Oxidation to the 3-acylpyrrol-2(5H)-one 36 was achieved by phenylselenation, and oxidative elimination using *m*-chloroperoxybenzoic acid at low temperature, in the presence of an excess of hydrogen peroxide, a procedure that effected selective oxidation of the selenide in the presence of the conjugated triene. The acylpyrrol-2-(5H)-one 36 was unstable and polymerized on attempted isolation, but could be detected in solution by ¹H NMR. Nevertheless, on heating in a dilute solution

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Scheme VI Synthesis of Proxiphomin (39)



at 100 °C, Diels-Alder cyclization occurred, to give two adducts, 37 and 38, 50-55%, ratio 52:48. N-Debenzoylation and phenylselenation-oxidative elimination. to introduce the 21,22-double bond, gave proxiphomin (39).21

It was reassuring that the Diels-Alder reaction of the trienoyl pyrrol-2(5H)-one 36 had been reasonably efficient and that it could be carried out under relatively mild conditions. This contrasted with the behavior of the analogous diene 40, which gave poor yields of Diels-Alder products on thermolysis, the lower reactivity of the conjugated diene allowing increased decomposition of the unstable 3-acylpyrrol-2(5H)-one to compete with cyclization.²⁰ However, the lack of exoendo selectivity observed during the cyclization of 36 was disappointing, although this appeared to be a feature of 13-membered-ring formation, since cyclization of 11-membered-ring precursors was much more stereoselective, vide infra.²⁰ No epimerization at C(5) of the pyrrol-2(5H)-one 36 was detected during its cyclization. A mixture of the 5S and 5R diastereoisomers 36 and 41 was synthesized from racemic pyrrolidinone and shown to give four adducts, i.e., 37 and 38 and the analogous adducts from the (5R)-pyrrol-2(5H)-one 41, which were not detected as products from $36.^{21}$

This intramolecular Diels-Alder strategy was applied to synthesize cytochalasin H (4).²² The chiral centers at C(16) and C(18) were introduced by stereoselective peracid epoxidation of the allylic alcohol 44, available from methyl (2S)-3-hydroxy-2-methylpropanoate (43), followed by reduction of the epoxide 45 using lithium aluminum hydride to give diol 46. Conventional chemistry was used to convert the diol into the long-chain aldehyde ester 47, which was converted into the conjugated triene 48, and into the 3-acylpyrrol-2(5H)-one 49 using the chemistry developed during the proxi-

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phomin synthesis. The Diels-Alder reaction of pyrrol-2(5H)-one 49 was carried out by heating a dilute solution in toluene at 80-100 °C for 5 h and was highly stereoselective, giving the required adduct 50 (38%) as the only isolable product. Mild base hydrolysis gave the N-deprotected lactam 51 (Scheme VII).

Modification of the 11-membered ring was carried out by phenylselenation-oxidative elimination, followed by reduction of the enone using sodium borohydride, this reduction being highly stereoselective due to the screening of the carbonyl group by the 11-membered ring. Acetylation gave allylic acetate 52, which was epoxidized to give a mixture of epoxides 53 and 54, although epoxidation of the ketone 51 gave only the C(6)-C(7) epoxide, perhaps because of through-bond electron withdrawal from the 13,14-double bond by the carbonyl group at C(21). Rearrangement of epoxide 54 gave the allylic alcohol 55, which was deprotected to provide cytochalasin H (4).²²

As an alternative procedure for introduction of the chiral centers at C(16) and C(18), citronellal (56) was converted via trienyl ester 57 and pyrrolidinone 58 into the 3-acylpyrrol-2(5H)-one 59. Diels-Alder cyclization was stereoselective and gave the adduct 60 (58%) together with a small amount, 1-2%, of its 13(14)-Z isomer. N-Deprotection and acetal hydrolysis gave the diketone 61, which reacted regio- and stereoselectively with methylmagnesium chloride in THF to give the tertiary alcohol 62 (84%), which has the same stereo-chemistry at C(18) as cytochalasin H (4)²³ (Scheme VIII).

Triene 57 was also incorporated into a synthesis of cytochalasin G (6).²⁴ The pyrrolidinone required for this synthesis, 63, was obtained from tryptophan ethyl ester and acylated by using the imidazolide derived from ester 57 to give the pyrrol-2(5H)-one 65 after oxidation. Diels-Alder cyclization gave 66, which was converted into cytochalasin G (6) by acetal hydrolysis,

Scheme VIII Alternative Approach to Cytochalasin H (4) and Synthesis of Cytochalasin G (6)



cyclohexene epoxidation, and N-deprotection.²⁴

Problems were encountered during attempts to use the intramolecular Diels-Alder approach to synthesize aspochalasin C (7).²⁵ Condensation of the dienyl phosphonate 31 with the methyl ketone 67, available from diethyl tartrate, gave the conjugated triene 68 as a mixture of 8E and 8Z isomers, ratio E:Z = 75:25. This mixture was not separated, but was converted to the imidazolide 69, which was used to acylate the lithium salt of the leucine-derived pyrrolidinone 70. Phenylselenation-oxidative elimination gave the pyrrol-2-(5H)-ones 72, which were heated as a dilute solution in toluene at 85 °C to effect Diels-Alder cyclization. Three Diels-Alder products were isolated, which were identified as the 13(14)-Z isomer 73 together with the two 13(14)-E isomers 74 and 75, ratio 3:1:2, respectively, combined yield 30%. It appeared that the major Diels-Alder product 73 had been derived from the minor (Z)-trienyl pyrrol-2(5H)-one (8'Z)-72, which had cyclized stereoselectively as usual, whereas the major (E)-trienyl pyrrol-2(5H)-one (8'E)-72 had cyclized nonstereoselectively to give a mixture of endo and exo isomers in which the usually observed endo product 74 was the *minor* component. On the basis of isolated vields, it was not clear whether E-Z isomerization of the trienes had been taking place under the reaction conditions, but the overall result was that the 13(14)-Eisomer 74 required for aspochalasan synthesis was only a minor product. The major product 73 was converted into the 13(14)-Z isomer 76 of aspochalasin C, by deprotection and phenylselenation-oxidative elimination²⁶ (Scheme IX).

The intramolecular Diels-Alder approach was finally applied to synthesize cytochalasin D (3).²⁶ Meth-

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Scheme IX Synthesis of Isoaspochalasin C (76)



acrolein was treated with (E)-2-butenyldiisopinocampheylborane (77), to give the anti homoallylic alcohol 78 after oxidative workup, together with a small amount (<10%) of its syn diastereoisomer. Claisen rearrangement using triethyl orthoacetate gave the ester 79, which was hydroborated and oxidized to give aldehyde 80. Condensation with the dienyl phosphonate 31 gave the conjugated triene 81, which as its imidazolide 82 was condensed with the pyrrolidinone 42. Oxidation by the usual procedure gave the 3-acylpyrrol-2(5H)-one 84, which on heating in dilute solution in toluene cyclized to give the Diels-Alder product 85 (25-35%), together with a minor diastereoisomer (ca. 4%) (Scheme X).

Whereas *m*-chloroperoxybenzoic acid reacted on the less hindered face of the 17,18-double bond of the adduct 85 to give epoxide 86, osmium tetraoxide reacted selectively on the 6,7-double bond to give diol 87 together with a small amount of the tetrol 88. Perhaps the peracid is reacting selectively at the more electron rich trisubstituted double bond, while osmium tetraoxide reacts at the more accessible.

Selective protection of diol 87, followed by dehydration, gave the exocyclic alkene 89, which was oxidized stereoselectively with osmium tetraoxide, to give acetonide 90 after diol protection. Phenylselenation, Ndeprotection, and oxidative elimination gave the enone 91, which was reduced with sodium borohydride-cerium(III) chloride and acetylated, followed by desilylation, SEM protection, and acetonide hydrolysis to give the diol 92. Finally, selective oxidation of the secondary alcohol at C(17) with oxalyl chloride-DMSO gave cytochalasin D (3) after deprotection.²⁶

During the course of this work, selective modifications based on tetrol 88 were investigated.²⁷ This tetrol could be isolated in an improved yield (69%) by using 2 molar equiv of osmium tetraoxide and was protected as its bis-acetonide 93. Introduction of the allylic acetate into the 11-membered ring was carried out as in the cytochalasin D synthesis, and selective hydrolysis of the 17,18-acetal gave diol 94. Swern oxidation of the C(17) secondary alcohol followed by acetal hydrolysis gave the trihydroxy ketone 95, which had spectroscopic data



identical with those of cytochalasin O,²⁸ so establishing the stereochemistry of this natural product at C(5) and C(6). An alternative approach to cytochalasin D has been reported.²⁹

Other Approaches to Cytochalasan Synthesis

Imides have been used as dienophiles during syntheses of the isoindolone fragment of the cytochalasans with regioselective reduction of the adducts 96 being developed to provide hydroxy lactams 97 ($\mathbf{R}' =$ H).³⁰ Treatment with acidic methanol then gave the ethers 97 ($\mathbf{R}' = \mathbf{M}\mathbf{e}$), which with tribenzylaluminum gave the isoindolones 98. Oxidation at C(9) gave 99, which may be useful for a cytochalasan B synthesis.³⁰ The intermolecular Diels-Alder reaction between the dienol 100 and the alkylidenemalonate 101 gave the adduct 102 with the wrong stereochemistry at C(9), but this was corrected by a ring-opening and -closing sequence to complete an efficient synthesis of the isoindolone 103.31,32 An alternative intramolecular Diels-Alder approach, developed in connection with a fragmentation route to the 11-membered ring, is based

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Scheme XI Other Isoindolone Syntheses



Scheme XII Synthesis of Cytochalasans via a Reformatski Reaction



on a Diels-Alder process using (Z)-dienes in which the C(4)-C(9) and C(7)-C(8) bonds are formed. For example, cyclization of the diene 104 gave a good yield of the isoindolone 105^{33,34} (Scheme XI).

Other procedures for the formation of the 11-membered ring of the carbocyclic cytochalasans include a modified Reformatski procedure³⁵ and a transitionmetal-mediated process.³⁶ Diels-Alder addition of the triene 106 with the 3-acylpyrrol-2(5H)-one 107 gave the isoindolone 108 (79%), which cyclized on slow addition of Rieke zinc in THF to give the cytochalasan 109 (75%) as a 1:1 mixture of epimers; samarium iodide was more selective but less efficient (46%). Dehydration of one of the isomers gave mainly the (19(20)-E)-alkene 110, which was taken through to the methylenecyclohexanol 111.35 The other epimer of 109 gave mainly the 19(20)-Z isomer of 110 on dehydration and led to a mixture of 111 and its 19(20)-Z diastereoisomer³⁵ (Scheme XII).

Aspochalasin B (117) has been synthesized via a transition-metal-mediated cyclization of the 11-membered ring.³⁶ Isoindolone 113 was prepared via an in-





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Scheme XIII Synthesis of Aspochalasin B (117)



Scheme XIV Synthesis of Zygosporin E (125) via Sulfur Ylide Chemistry



termolecular Diels-Alder reaction between diene 100 and the leucine-derived dienophile 112 and was converted into aldehyde 114, with the correct configuration at C(9), by the ring-opening and -closing sequence. Modification of the ester and aldehyde groups gave the sulfone 115, which was cyclized by using a palladium catalyst, to give the aspochalasan precursor 116 as a single diastereoisomer (49%). Stereoselective epoxidation of the enol ether double bond followed by hydrolysis and elimination of the sulfone gave aspochalasin B (117)³⁶ (Scheme XIII).

In addition to the direct formation of the 11-membered ring of the cytochalasans, ring-expansion and ring-fragmentation procedures have been developed, one useful route being based upon the 2,3-rearrangement of sulfonium ylides.^{37,38} The intermolecular Diels-Alder reaction between the triene 118 and the acylpyrrol-2(5H)-one 107 gave adduct 119 (93%). This was used to generate the thioaldehyde 120, which was trapped by a 2-silylated butadiene to give the dihydrothiopyran 121 as the major product after equili-

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bration under basic conditions. Conversion into the thiopyranone 122 followed by treatment with sodium iodide generated the sulfonium ylide 123, which rearranged to form the sulfur-bridged cytochalasan 124 (65%) together with minor isomeric products.³⁸ Further modification gave rise to zygosporin E (125)³⁹ (Scheme XIV).

Another approach to the 11-membered-ring system of the cytochalasans based on fragmentation of keto toluene-p-sulfonates has been described.⁴⁰

Conclusion

Intramolecular Diels-Alder reactions of 3-acylpyrrol-2(5H)-ones would appear to provide useful access to macrocyclic compounds including [11]cytochalasans, although improved methods are required for stereose-

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lective 13-membered-ring formation. Conformational factors affecting the endo-exo selectivities of these reactions have not been elucidated, although it is likely that they are under kinetic control. Further work on other approaches to cytochalasan synthesis is expected, thus making these complex, biologically active molecules, and their analogues, more readily available for chemical and biological study. Further insights into the processes involved in their biosynthesis are of interest. Is an enzymically mediated Diels-Alder reaction involved?41,42

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Calicheamicins: Discovery, Structure, Chemistry, and **Interaction with DNA**

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Introduction

The enediyne class of antitumor antibiotics represented in Chart I by calicheamicin $\gamma_1^{I}(1)$,¹ esperamicin $A_1(2)$,² dynemicin A (3),³ and neocarzinostatin (4)⁴ are some of the most potent antitumor agents ever discovered. Calicheamicin γ_1^{I} is over 1000 times more potent than adriamycin, a clinically useful antitumor antibiotic, when tested in murine tumor models. The remarkable biological properties of this class of compounds appear to be a consequence of their ability to interact with cellular DNA and initiate double-stranded

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cleavage by carbon-centered diradical hydrogen abstraction processes. The elucidation of the chemical structures of 1 and 2 has stimulated intense synthetic efforts in a number of academic laboratories.

Similarities between the calicheamicins and the esperamicins were recognized quite early in our studies of the calicheamicins due to the extreme potency of both families of compounds, the presence of a thio sugar in both, and characteristic proton signals of their aglycons. However, the close structural relationships between these two families of compounds were not recognized until the structures of calicheamicin γ_1^{I} and the esperimicins were published simultaneously.^{1,2} Dynemicin A shares with the calicheamicins in having a 10-membered enediyne ring system, although the remainder of its structure is very different. The chemical structure of the neocarzinostatin chromophore is only remotely related to that of the calicheamicins. The conjugated diacetylenic ring systems of these potent antitumor antibiotics can undergo cycloaromatization

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